Question 21 (3 marks)	
A $0.001 \text{ mol } L^{-1}$ solution of hydrochloric acid and a $0.056 \text{ mol } L^{-1}$ solution of ethanoic acid both have a pH of 3.0.	3
Why do both solutions have the same pH?	
trend both have the same on bootuse	
mey are for acional substances	
The state of the s	