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Question 24 (4 marks)

In the margarine industry, alkenes are often hydrogenated to convert unsaturated oils into solid fats that have a greater proportion of saturated molecules.

Using ethene as an example, write an equation for this reaction and state the type of reaction this represents.	2
$C_2H_4 + H_2O \longrightarrow C_2H_5OH$	
.this is an example of a hydration reaction.	
Describe a test that could be used to confirm that all the ethene has been converted.	2
pour some of the product into a bowl	
and light it. If it built it all then its all been	
concerted to ethonol if not then there is still some	
	of reaction this represents. $C_2 H_4 + H_2 O \longrightarrow C_2 H_5 O H$ this is an example of a hydration reaction. Describe a test that could be used to confirm that all the ethene has been converted. $main cont (20mi)pour Some of the product into a bows and sign it. If it burns it all then its all been$

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