Question 24 (4 marks)

In the margarine industry, alkenes are often hydrogenated to convert unsaturated oils into solid fats that have a greater proportion of saturated molecules.

(a) Using ethene as an example, write an equation for this reaction and state the type 2 of reaction this represents.

 $\frac{C_2H_4}{\omega} + \frac{H_2}{\omega} \rightarrow \frac{C_2H_6}{\omega}$

this reaction represents a hydration and redar reaction.

(b) Describe a test that could be used to confirm that all the ethene has been 2 converted.

A test that could be used is the brownine water test. The brownine water acts as an indicator indicating whether it is an alkene or alkane. The brownine water will torn colour in the Presense of an allere. However, brownine water will not change colour if it is an alkane. Therefore by carrying this test, it can be confirmed if all the ethene has been converted ethane, if there is no colour change.

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