8

Question 28 (8 marks)

The flowchart shown outlines the sequence of steps used to determine the concentration of an unknown hydrochloric acid solution.

Preparation of A
500 mL 0.100 mol L⁻¹ sodium
carbonate standard solution

25.0 mL used

Unknown
hydrochloric acid
solution

Average titration
volume of acid 21.4 mL

Concentration of hydrochloric acid solution

Describe steps **A**, **B** and **C** including correct techniques, equipment and appropriate calculations. Determine the concentration of the hydrochloric acid.

Hu right liquid burette (acid) flash

pipette base clinical flash (natur)

Add the liquids in and let the

burette drip ite the clinical

flash with the base solution

of the pipette minth on

Question 28 continues on page 18

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